

CHUCHMAREV, S.K.

256T82

USSR/Metallurgy - Steel, Isotopic
Analysis

Jan 53

"Determination of the Distribution Rate of a
Substance in Liquid Metal (Method of Radio-
active Indicators)," M.M. Karnaukhov, Corr
Mem Acad USSR, S.K. Chuchmarev

Iz Ak Nauk SSSR, OTN, No 1, pp 82-95

Develops method for lab study of processes con-
nected with transfer of substance in liquid
metal without sampling, using radioactive sub-
stances capable of emission of gamma rays as
indicators.

256T82

Studies diffusion of Co and Fe in liquid iron
at temp of $1,620 \pm 20^\circ$, using indicators Co^{60}
and Fe^{59} and in eutectic Fe-C alloy.

~~CHUCHMAREV, S.K.~~

24-9-17/33

AUTHORS: Barmin, L. N., Yesin, O.A. and Chuchmarev, S.K.(Sverdlovsk)

TITLE: Study by the e.m.f. method of the properties of hydrogen which is dissolved in liquid slags. (Izucheniye svoystv vodoroda, rastvorennogo v zhidkikh shlakakh, metodom elektrodvishushchikh sil)

PERIODICAL: Izvestiya Akademii Nauk SSSR, Otdeleniye Tekhnicheskikh Nauk, 1957, No.9, pp.114-118 (USSR)

ABSTRACT: For accurate determination of small quantities of H_2 dissolved in slag it is of interest to use the e.m.f. method which permits study of the behaviour of the hydrogen directly inside the liquid slags; for this purpose it is necessary to build a galvanic cell with oxygen and hydrogen electrodes. The authors of this paper considered it advisable to investigate the behaviour of a hydrogen electrode in slags and to compare two types of oxygen electrodes, namely, the gaseous one and the one made of solid magnesium oxides. Three types of circuits were studied, namely, the hydrogen, the hydrogen-oxygen with a barrier made of MgO and the hydrogen-oxygen with one slag. The experiments were carried out in a SiC furnace, the temperature being measured with a Pt-Pt Rh thermocouple. The diagram of the cell for the first two mentioned circuits

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Study by the e.m.f. method of the properties of hydrogen which is dissolved in liquid slags.

is that shown in Fig.1, the diagram of the last mentioned circuit is shown in Fig.2. Data and the results are entered in tables and plotted in graphs. The authors have proved experimentally that the reversible hydrogen electrode can be materialised relatively simply for molten slags which do not contain easily reducible oxides. New experimental data confirm that solid magnesium oxide which is in contact with the liquid slag operates as a sort of oxygen electrode. The measurements have shown that the activity of the water dissolved in the slag is proportional to the square root of the activity of the calcium oxide; this is in agreement with the assumption that the hydrogen in the slag is present in the form of hydroxyl anions. The relation between the e.m.f. of the oxygen-hydrogen cell and the oxygen activity permits considering the negative magnitude of the e.m.f. as a measure of the basicity of the slag. There are 4 figures, 5 tables and 15 references, 5 of which are Slavic.

Card 2/2

SUBMITTED: May 20, 1957.

ASSOCIATION: Ural Polytechnical Institute (Ural'skiy Politekhni-cheskiy Institut), Sverdlovsk

AVAILABLE: Library of Congress.

CHUCHMAREV, S. K.

✓ The mutual solubility of calcium oxide and calcium carbonate. P. V. Gel'd, A. I. Pashlov, and S. K. Chuchmarev (S. M. Khov Ural Polytech. Inst.). *Doklady Akad. Nauk S.S.S.R.* 91, 1115-17(1953).—The mutual sol. of CaCO_3 and CaO was studied at $\sim 17^\circ$ by the x-ray structural method (i.e., the precise detn. of lattice parameters of the pure components and of the products from the partial decompn. of the carbonate). The wt. % of CaO was varied from 0 to 100. The exptl. results show that CaCO_3 is only negligibly sol. in CaO . A soln. of 1% CaO in CaCO_3 changes the parameters of the latter. J. Rovtar Leach

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(2)

CHUCHMAREV, S.K., YESIN, O.A., BARMIN, L.N.

"Gas Permeability of Liquid Slags,"
lecture given at the Fourth Conference on Steelmaking, A.A. Baikov Institute of
Metallurgy, Moscow, July 1-6, 1957

YESIN, O.A.; VORONTSOV, Ye.S.; GRUCHMAREV, S.K.

Diffusion of phosphorus and calcium in the fusions $\text{CaO} - \text{Al}_2\text{O}_3 -$

SiO_2 and $\text{CaO} - \text{P}_2\text{O}_5$. O.A. Zhur. fiz. khim. 31 no.10:2322-2327

0 '57.

(MIRA 11:3)

1. Ural'skiy politekhnicheskiy institut im. S.M. Kirov a.
(Diffusion) (Phosphorus) (Calcium)

CHUCHMAREV, S.K., kand.tekhn.nauk, dotsent; YESIN, O.A., doktor tekhn.nauk,
prof.; BARMIN, L.N., inzh.

Effect of electric current on the behavior of hydrogen dissolved
in liquid metal. Izv. vys. ucheb. zav.; Chern.Met. no.5:59-64
My '58.

(MIRA 11:7)

1. Ural'skiy politekhnicheskiy institut.
(Metals--Hydrogen content) (Liquid metals)

BARMIN, L.N., inzh.; YESIN, O.A., doktor tekhn.nauk, prof.; CHUCHMAREV,
S.K., kand.tekhn.nauk, dotsent

Effect of slag composition on the activity of the hydrogen dissolved in it. Izv.vys.ucheb.zav.; chern.met. no.6:65-73 Je '58.
(MIRA 12:8)

1. Ural'skiy politekhnicheskiy institut. Rekomendovano kafedroy teorii metallurgicheskikh protsessov Ural'skogo politekhnicheskogo instituta.

(Slag—Analysis) (Hydrogen) (Activity coefficients)

BARMIN, L.N.; YESIN, O.A.; CHUCHMAREV, S.K.

Determining water activity in slag by electrochemical methods.
Trudy Ural. politekh. inst. no.93:28-38 '59. (MIRA 15:3)
(Slag) (Water) (Electromotive force)

CHUCHMAREV, S.K.; BARMIN, L.N.

Effect of slag basicity on its penetrability to gas. Trudy Ural.
politekh. inst. no.93:39-43 '59. (MIRA 15:3)
(Slag) (Hydrogen-ion concentration)

YESIN, O.A.; POPEL, S.I.; CHUCHMAREV, S.K.

Sulfur removal from slag by electrolysis. Izv.vys.ucheb.zav.;
chern.met. no.3:5-9 '60. (MIRA 13:4)

1. Ural'skiy politekhnicheskiy institut.
(Slag) (Desulfuration)

NOVOKHATSKIY, I.A.; YESIN, O.A.; CHUCHMAREV, S.K.

Methods of determining the diffusion coefficient of hydrogen in
molten slags. Izv.vys.ucheb.zav.; chern.met. no.4:5-14 '61.
(MIRA 14:4)

1. Ural'skiy politekhnicheskii institut.
(Slag) (Activity coefficients) (Hydrogen)

CHUCHMAREV, S.K.; YESIN, O.A.; BARMIN, L.N.

Cathodic behavior of hydrogen dissolved in molten oxides. Izv. vys.
ucheb. zav.; chern. met. 4 no.8:9-17 '61. (MIRA 14:9)

1. Ural'skiy politekhnicheskiy institut.
(Slag) (Hydrogen-ion concentration)

NOVOKHATSKIY, I.A.; YESIN, O.A.; CHUCHMAREV, S.K.

Mechanism of hydrogen diffusion in slags. Izv. vys. ucheb. zav.;
chern. met. 4 no.10:10-18 '61. (MIRA 14:11)

1. Ural'skiy politekhnicheskiy institut.
(Diffusion) (Slag)

NOVOKHATSKIY, I.A.; YESIN, O.A.; CHUCHMAREV, S.K.

Hydrogen solubility in molten slags. Izv. vys. ucheb. zav.; chern.
met. 4 no.11:22-29 '61. (MIRA 14:12)

1. Ural'skiy politekhnicheskiy institut.
(Slag) (Hydrogen)

CHUCHMAREV, S. K.
NOVOHATSHI, I. A. [Novokhatskiy, I. A.]; ESIN, O. A. [Yesin, O. A.];
CHUCHMAREV, S. K. [Chuchmarev, S. K.]

A method of determining the diffusion coefficient of hydrogen in
melted slag. *Analele metalurgie* 15 no.4:5-15 Q-D '61.

(Slag) (Hydrogen) (Diffusion)

S/020/61/136/004/022/026
B028/B060

AUTHORS: Novokhatskiy, I. A., Yesin, O. A., and Chuchmarev, S. K.
TITLE: Diffusion of Hydrogen in Molten Slag
PERIODICAL: Doklady Akademii nauk SSSR, 1961, Vol. 136, No. 4,
pp. 868-870

TEXT: Data available in the literature concerning the mass transfer of hydrogen in molten slag indicate very high values (10^{-3} - 10^{-2} cm² sec⁻¹). These data were obtained under the conditions prevailing in open-hearth furnaces, and rather stand for convection than for molecular diffusion D_H . To eliminate convection entirely and to approach the value for D_H , a nonsteady diffusion was used in the present work. An Al₂O₃ test tube contained a thin layer of viscous, liquid slag ($\delta = 1.5$ mm, $\eta = 3 - 100$ poises, 1410 - 1608°C) of the composition 16.5 - 53.0% CaO, 8.2 - 41.0% Al₂O₃, 6.0 - 58.3% SiO₂. Dried nitrogen was blown through to convey the water liberated from the slag to a hygrometer. The dew point was used to calculate the rate of water yielded by the slag. ✓

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Diffusion of Hydrogen in Molten Slag

S/020/61/136/004/022/026
B028/B060

$Q_t/Q_0 \approx 1 - 8/\pi^2 e^{-\theta}$ (1). If θ is known, it is possible to calculate the diffusion coefficient. $D_H = \frac{4\delta^2\theta}{\pi^2\tau}$ (2). The values found for D_H assuming

three thicknesses of the slag layer (1.3; 1.8; and 2.6 mm) proved to be very high (1.0; 1.1; 0.9) $\cdot 10^{-5}$ cm²/sec. V_{H_2O} was not dependent upon the layer thickness. The fact that convection played no role in the experiment was checked with Lin' Tszya-tszyao (Ref. 3) and confirmed. The D_H found thus characterized the molecular diffusion of H₂ which was assumed to diffuse through the slag in the form of protons, and to pass over from one oxygen atom to another. This transition takes place only if the distance to the adjacent O atom does not exceed 2.65 Å, as occurs with SiO₂ (d = 2.64 Å). D_H was practically constant in slag with 56.4% SiO₂. D_H rises with an increase of CaO and so does the activation energy. Due to

$$D = 2.72 \frac{kT}{h} \lambda^2 \exp(\Delta S^*/R) \exp\left(\frac{-E}{RT}\right). (\Delta S^* = \text{activation entropy, } \lambda \text{ distance})$$

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Diffusion of Hydrogen in Molten Slag

S/020/61/136/004/022/026
B028/B060

between the equilibrium states of the moving particle), D_H and E may then increase at the same time, if λ rises. λ rises in CaO (lattice distance of the O atoms = 3.41 Å). In the case of slag rich in CaO, D_H amounted to $2.05 \cdot 10^{-5}$ cm²/sec, E = 20800 cal/mole. There are 3 figures and 16 references: 10 Soviet, 4 US, 1 Canadian, and 1 British.

ASSOCIATION: Ural'skiy politekhnicheskii institut im. S. M. Kirova, Sverdlovsk (Ural Polytechnic Institute imeni S. M. Kirov, Sverdlovsk) ✓

PRESENTED: July 20, 1960, by A. N. Frumkin, Academician

SUBMITTED: July 9, 1960

Card 3/3

CHUCHMAREV, S.K.; YESIN, O.A.; DOBRYDEN', A.A.

Oxidation kinetics of slag sulfur by gaseous oxygen. Izv. vys.
ucheb. zav.; Chern. met. 5 no.7:12-18 '62. (MIRA 15:8)

1. Ural'skiy politekhnicheskiy institut.
(Sulfur) (Oxidation)

CHUCHMAREV, S.K.; YESIN, O.A.; DOBRYDEN', A.A.

Effect of the electrode polarization on the properties of the
electrolyte-gas boundary. Dokl. AN SSSR 144 no.5:1100-1102
Je '62. (MIRA 15:6)

1. Ural'skiy politekhnicheskiy institut imeni S.M.Kirova.
Predstavleno akademikom A.N.Frumkinym.
(Sulfur) (Combustion) (Polarization (Electricity))

CHUCHMAREV, S.K.; YESIN, O.A.; NOVOKHATSKIY, I.A.

Hydrogen permeability through molten slags. Izv. vys. ucheb.
zav.; Chern. met. 5 no.10:5-13 '62. (MIRA 15:11)

1. Ural'skiy politekhnicheskiy institut.
(Slag) (Hydrogen)

DOBRYDEN', A. A.; YESIN, O. A.; CHUCHMAREV, S. K.

Kinetics of the burning out of sulfur in slags during the
passage of a direct electric current. Izv. vys. ucheb. zav.;
chern. met. 5 no.12:12-19 '62. (MIRA 16:1)

1. Ural'skiy politekhnicheskiy institut.

(Slag—Sulfur content) (Electrolysis)

CHUCHMAKOV, S.K.; DOBRYDEN', A.A.

Using electrical current to intensify the burning out of
sulfur in slags. Spets. nauch. trud. Ural. politekh. inst.
no.126:34-48 '63 (MIRA 17:8)

DOBRYDEN', A.A.; YESIN, O.A.; CHUCHMAREV, S.K.

Effect of alternating current on the rate of the burning-out of
sulfur from slag. Izv. vys. ucheb. zav.; chern. met. 6 no.3:
20-25 '63. (MIRA 16:5)

1. Ural'skiy politekhnicheskiy institut.
(Desulfuration) (Electric currents, Alternating)

CHUCHMAREV, S.K.; YESIN, O.A.

Form of hydrogen existence in molten slags. Izv. vys. ucheb. zav.;
chern. met. 6 no.4:12-19 '63. (MIRA 16:5)

1. Ural'skiy politekhnicheskiy institut.
(Slag-Testing) (Hydrogen ion concentration)

CHUCHMAREV, S.K.; YESIN, O.A.; PASTUKHOV, E.A.

Form in which trivalent iron ions occur in molten silicates. Dokl.
AN SSSR 150 no.5:1094-1096 Je '63. (MIRA 16:8)

1. Ural'skiy politekhnicheskiy institut im. S.M.Kirova.
Predstavleno akademikom A.N.Frumkinym.
(Iron compounds) (Silicates)

LOBRYDEN', A. A.; YESIN, O. A.; CHUCHMAREV, S. K.

Intensifying the desulfuration of cast iron by the electrolysis
of slag. Izv. vys. ucheb. zav.; chern. met. 7 no.6:11-16 '64.
(MIRA 17:7)

1. Ural'skiy politekhnicheskii institut.

DOBRYDEN', A.A.; YESIN, O.A.; CHUCHMAREV, S.K.

Kinetics of the desulfuration of pig iron and slag by direct
and alternating currents. Izv. vys. ucheb. zav.; Chern. met. 7
no.2:5-12 '64. (MIRA 17:3)

1. Ural'skiy politekhnicheskiy institut.

KAMYSHOV, V.M.; YESIN, O.A.; CHUCHMAREV, S.K.

Nitrogen solubility in iron-free slags. Izv. vys. ucheb. zav.;
chern. met. 7 no.7:24-28 '64 (MIRA 17:8)

1. Ural'skiy politekhnicheskiy institut.

CHUCHMAREV, S.K.; YESIN, O.A.; KAMYSHOV, V.M.; DOBRYDEN', A.A.

Kinetics of nitrogen dissolution in fused iron-free slags.

Izv. vys. ucheb. zav.; chern. met. 7 no.9:11-15 '64.

(MIRA 17:6)

1. Ural'skiy politekhnicheskiy institut.

DOBRYDEN', A.A.; YESIN, O.A.; CHUCHMAREV, S.K.

Electrochemical desulfuration of varying basicity slags.

Izv. vys. ucheb. zav.; Chern. met. 7 no.11:21-23 '64.

(MIRA 17:12)

1. Ural'skiy politekhnicheskiy institut.

CHUCHMAREV, S.K.; KAMYSHOV, V.M.

Chizhevskii method applied to determine nitrogen in slags. Zav.
lab. 30 no.9:1068-1069 '64. (MIRA 18:3)

1. Ural'skiy politekhnicheskii institut imeni Kirova.

PASTUKHOV, E.A.; YESIN, O.A.; CHUCHMAREV, S.K.

Particular characteristics of the diffusion of iron ions in molten aluminosilicates. Elektrokimiia 1 no.1:78-83 Ja '65. (MIRA 18:5)

1. Ural'skiy politekhnicheskii institut im. S.M. Kirova.

PASTUKHOV, E.A.; YESIN, O.A.; CHUCHMAREV, S.K.

Form of ions present in molten slags. Zhur. fiz. khim. 38
no.5:1306-1310 My '64. (MIRA 18:12)

1. Ural'skiy politekhnicheskii institut. Submitted June 8, 1963.

KAMYSHOV, V.M.; YESIN, O.A.; CHUCHMAREV, S.K.; DOBRYDEN', A.A.

Effect of the electric current on the rate of nitrogen dissolution
in molten oxides. Elektrokimiia 1 no.2:227-230 F '65.

(MIRA 18:6)

1. Ural'skiy politekhnicheskii institut imeni Kirova.

CHUCHMAREV, S.K.; YESIN, O.A.; KAMYSHOV, V.M.

Form of nitrogen existing in molten nonferrous slags. Izv. vys.
ucheb. zav.; Chern. met. 8 no.2:5-9 '65.

(MIRA 18:2)

1. Ural'skiy politekhnicheskii institut.

TIKHONOV, A.I.; CHUCHMAREV, S.K.; SMIRNOV, V.I., akademik

Kinetic regularities in the oxidation of lower nickel sulfide in a fluidized bed. Dokl. AN SSSR 163 no.3:686-689 J1 '65. (MIRA 18:7)

1. Ural'skiy politekhnicheskiy institut im. S.M.Kirova. 2. AN KazSSR (for Smirnov).

PASTUKHOV, E.A. (Sverdlovsk); YESIN, O.A. (Sverdlovsk); CHUCHMAREV, S.K.
(Sverdlovsk)

Kinetics of the oxidation of divalent iron in slag by gaseous
oxygen. Izv. AN SSSR. Met. no.4:51-56 J1-Ag '65.

(MIRA 18:8)

L 16133-66

EPF(n)-2/EWP(k)/EWT(m)/ETC(f)/EWG(m)/EHA(d)/EWP(e)/EWP(t)
ACC NR: AP6004185 IJP(c)
WH/WW/JD/JG/WB SOURCE CODE: UR/0076/66/040/001/0262/0263

AUTHOR: Kamyshov, V. M.; Yesin, O. A.; Chuchmarev, S. K.

ORG: Ural Polytechnic Institute im. S. M. Kirov (Ural'skiy politekhnicheskiy institut)

TITLE: Wetting of transition metal nitrides by molten oxides and metals

SOURCE: Zhurnal fizicheskoy khimii, v. 40, no. 1, 1966, 262-263

TOPIC TAGS: nitride, transition metal, copper, iron, calcium oxide, aluminum oxide, silicon dioxide, carbon alloy, surface tension, wetting

ABSTRACT: The contact angle θ of wetting of transition metal (Ti, V, Nb, Cr, Mo, Zr) nitrides by copper^{44.55} iron (99.9% Fe), iron-carbon alloy (0.5% C), ShKh-15 steel, and molten oxides (I - 50% CaO, 50% Al₂O₃; II - 40% CaO, 40% SiO₂, 20% Al₂O₃) was measured at 1500 - 1550C by the sessile drop method. The contact angle of wetting by copper of solid transition metal nitrides was found to be large and independent of the nature of the nitride. On the contrary, in the case of wetting by iron, steel, and oxide

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UDC: 541.17

L 16133-66

ACC NR: AP8004185

melts, θ decreases with increasing acceptor capacity criterion $1/Nn$, where n is the number of electrons in the d subshell and N is the principal quantum number of the d -subshell level. As the different nitrides are considered, differences observed in the change of θ from one case to another are due to both the surface tension and the interfacial tension at the solid-liquid interface. Orig. art. has: 1 figure and 1 formula.

SUB CODE: 07/1 SUBM DATE: 10May65/ ORIG REF: 003

Liquid metal ¹⁸

Card 2/2

CHUCILLO, E.

Shielded arc welding of thin steel sheets. p. 246.
PRZEGLAD SPAWALNICTWA, Warszawa, Vol. 6, no. 11, Nov. 1954.

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 4, no. 10, Oct. 1955,
Uncl.

POL. 1

0127 021.791.75 : 021.791.54 : 020.179.3
Chuchro E. Powders and Wire Electrodes for Automatic Covered-Arc
Welding

"Proszki i druty do automatycznego spawania łukiem 'zrytym'".
Przegląd Spawalnictwa, No. 8, 1954, pp. 172-174.

The Welding Institute has developed two kinds of powders and two kinds of wire electrodes for the automatic welding of low-carbon steels by covered arc technique. The powders consist chiefly of compound aluminum silicides of magnesium and calcium, together with small quantities of calcium fluoride and alkalis. The granulation of powders for welding sheets not exceeding 8 to 15 mm amounts to

less than 1.5 mm, but finer powders should be used for welding thicker sheets. Two kinds of wire electrodes containing from 1.5 to 1.8 percent manganese were developed for two varieties of fluxes of a varying degree of basicity ($\text{CaO} + \text{MgO} : \text{SiO}_2$). A concise outline is given of the technology of flux production.

CHUCHUKALO, A.I.

Change in the amount of lipids, cholesterol, and ascorbic acid
in the adrenal cortex of rats after lesions due to polonium and
X irradiation. Med. rad. 5 no.8:37-41 '60. (MIRA 13:12)
(RADIATION SICKNESS) (ADRENAL CORTEX)

COUNTRY	:USSR
CATEGORY	:General Problems of Pathology. Inflammation
ABST. JOUR.	: RZBiol., No. 12 1958, No. 56175
AUTHOR	: <u>Chuchukalo, A. A.</u>
INST.	:-
TITLE	:The Influence of Inflammation on the Phagocytic Reaction of the Reticulo-Endothelial System following Polonium Injury of Animals
ORIG. PUB.	:Tr. Vses. Konfrentsii po Med. radiol. Ekspirim. Med. radiol. Moscow, Medgiz, 1957, 174-178
ABSTRACT	:Mice were injected subcutaneously with polonium (P; 0.005 microcuries each). Inflammation was induced by the subcutaneous injection of turpentine. The phagocytic activity (PA) of the RE system was appraised by observations of the uptake by liver RE cells of intraperitoneally-injected 1% solution of trypan blue or of India ink at 1, 5, and 10 days after the injection of P or of turpentine. The PA increased with the development of inflammation. With injury due to P alone, the PA was inhibited for the first five days and on the 10th to 15th days, but was above normal on the 5th to 10th
CARD:	1/2

CATEGORY :

ABS. JOUR. : RZBiol., No. 1958, No.

AUTHOR :

FILE :

PRICE :

ORIG. PUB. :

ABSTRACT : days. Comparable changes were seen in the PA upon simultaneous injection of P and turpentine, and also upon the development of inflammation 3 days after the onset of radiation. With injury on a background of 3-day inflammation, the PA occurred at a more rapid rate than in the case of P injury alone, but somewhat less markedly than with inflammation alone. --L.A.Oyvin

CARD: 2/2

AUTHORS: Berezin, I. V., Makalets, B. I., SOV/79-28-10-19/60
Chuchukina, L. G.

TITLE: Mechanism of the Oxidation of the Acids With Molecular Oxygen
in the Medium of n-Heptane (Mekhanizm okisleniya kislot
molekulyarnym kislородom v srede n-heptana)

PERIODICAL: Zhurnal obshchey khimii, 1958, Vol 28, Nr 10, pp 2718-2723
(USSR)

ABSTRACT: From the papers known on the oxidation of acids with bound
and air oxygen in the presence of catalysts (Refs 1-4) it may
be seen that the oxidation mechanism of the acids depends on
the conditions of the experiments. Therefore the rules
governing the oxidation of the single acids in the presence
of catalysts with different oxidizing agents may not be
extended without earlier examination to the case where the
oxidation of the acids takes place in the medium of an
oxidizing hydrocarbon. It was of interest to the authors to
investigate the chemical nature of the oxidation of acids in
this respect, to compare it with data in publications and thus
to discover the fundamentals of the oxidation mechanism in
dependence on the character of the reaction and the structure

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Mechanism of the Oxidation of the Acids With
Molecular Oxygen

SOV/79-28-10-19/60

of the acid molecules. As the acids are not only used up in the reaction process but also are formed as such as a consequence of the oxidation of hydrocarbon, the favorable solution of this problem consists of employing the method of isotopic indicators. A simple method was chosen that made the analytical part of the work considerably easier, i. e. the oxidation of n-butyric and n-valeric acid in n-heptane medium. To observe the behaviour of the functional group as well as that of the hydrocarbon chain of the acid an n-butyric acid with radioactivated carbon in the carboxyl, and an n-valeric acid radioactivated in the α -position were synthesized. Moreover, an acetic acid was produced that was radioactivated in the carboxyl in order to prove its oxidizability under the conditions given. Concluding, the following results are mentioned: The acids are subjected to a quantitative decarboxylation in the medium of the oxidizing hydrocarbon. In the activation of the carboxyl with radioactivated carbon CO_2 is the only active gaseous reaction product. According to this separated gas the behaviour of the acid carboxyl in any complex system of the oxidation products of hydrocarbons can be classified. The oxidizing reagent attacks the acid/molecule

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Mechanism of the Oxidation of the Acids With
Molecular Oxygen

SOV/79-28-10-19/60

at the β -carbon atom. In the decarboxylation of the acid a methyl ketone is formed that has one carbon atom less than the acid. The oxidizability of the acid depends on its structure. The acetic acid is practically inert. The yield of the n-valeric acid activated with radioactivated carbon in the α -position amounted to 23 %. There are 1 table and 9 references, 3 of which are Soviet.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet
(Moscow State University)

SUBMITTED: August 8, 1957

Card 3/3

CHUCHUKINA L.G.

~~Oxidation of Hydrocarbons in the~~ (Cont.)

SOV/3663

Knorre, D.G., L.G. Chuchukina, and N.M. Emanuel' [Institute of Chemical Physics]. Dual Function of Metal Stearates in the Hydrocarbon Oxidation Reaction

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The dual role of copper and manganese stearates as both catalysts and inhibitors of oxidation of iso- and n-decanes is described. The authors determine the critical concentration of cupric stearate ($\sim 0.03\%$ per mole) above which the induction period for n-decane oxidation increases.

Mayzus, Z.K., L.G. Privalova, and N.M. Emanuel' [Institute of Chemical Physics]. Change in the Mechanism of n-Decane Oxidation in the Course of the Reaction

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The authors have used C^{14} tagged n-decane to investigate changes in the rates of formation and consumption of n-decyl hydroperoxides during the oxidation of n-decane. The hypothesis that variations in the activities of radicals carrying on chain reactions are proportional to the accumulation of oxygen-containing oxidation products in the reacting mixture is offered as a possible explanation of the phenomenon.

Oxidation of Hydrocarbons in the Liquid Phase; Collection of Articles, Moscow, Izd-vo AN SSSR, 1959, 334pp. (Ak. nauk SSSR, Inst. khim. fiziki)

5(4)

AUTHORS:

Knorre, D.G., Chuchukina, L. G.,
Emanuel', N.M.

SOV/76-33-4-20/32

TITLE:

On the Phenomenon of Critical Concentration of $\text{Cu}(\text{C}_{17}\text{H}_{35}\text{COO})_2$
in the Reaction of Catalytic Oxidation of n-Decane
(O yavlenii kriticheskoy kontsentratsii $\text{Cu}(\text{C}_{17}\text{H}_{35}\text{COO})_2$ v
reaktsii katalizirovannogo okisleniya n-dekana)

PERIODICAL:

Zhurnal fizicheskoy khimii, 1959, Vol 33, Nr 4, pp 877 - 882
(USSR)

ABSTRACT:

In the present paper a new example is given of the critical phenomena in reactions with degenerated branchings in liquid phase which were observed in the investigations of n-decane oxidation in the presence of copper stearate (I). At a certain concentration of (I) a complete stop of the reaction was observed while at this concentration a self-accelerated reaction with a small induction period takes place. n-decane (II) was produced electrolytically (Ref 5) and (I) according to a method of production described. (I) and (II) were dissolved in a nitrogen current and the moment where oxygen was introduced was regarded as the beginning of reaction.

Card 1/3

On the Phenomenon of Critical Concentration SOV/76-33-4-20/32
of $\text{Cu}(\text{C}_{17}\text{H}_{35}\text{COO})_2$ in the Reaction of Catalytic Oxidation of n-Decane

0.03 - 0.10 mol% of (I) were used and the peroxides, carbonyl compounds, acids, and copper were determined according to the course of oxidation; the latter according to two methods:

Cu^{2+} and the entire copper. A catalytic effect (Fig 1 for 0.03% (I)) could be observed until a concentration of 0.06 mol% (I) is attained. A change in the copper valency in the induction period is explained by a reaction of (I) with intermediate oxidation products of (II) (e.g. hydrogen peroxides) which causes the self-acceleration proper of the process. The increase of the induction period at an increase of concentration of (I) is indicative of a second- the inhibiting - effect of (I) which apparently is based on a destruction of the chains at the (I)-molecules. The critical concentration is at 0.065 mol% (I) at which the induction period is longer than 15 hours and where the oxidation rate also changes. The phenomenon of the critical concentration of (I) is explained by the radical-chain-mechanism of hydrocarbon oxidation. There are 5 figures, 1 table, and 9 references, 6 of which are Soviet.

Card 2/3

On the Phenomenon of a Critical Concentration SOV/76-33-4-20/32
of $\text{Cu}(\text{C}_{17}\text{H}_{35}\text{COO})_2$ in the Reaction of Catalytic Oxidation of n-Decane

ASSOCIATION: Akademiya nauk SSSR, Institut khimicheskoy fiziki, Moskva
(Academy of Sciences of the USSR, Institute of Chemical
Physics, Moscow)

SUBMITTED: September, 28, 1957

Card 3/3

S/081/61/000/002/002/023
A005/A105

Translation from: Referativnyy zhurnal, Khimiya, 1961, No. 2, p. 59, # 2B437

AUTHORS: Knorre, D.G., Chuchukina, L.G., Emanuel', N.M.

TITLE: On the Double Function of Metal Stearates in the Oxidation Reaction of Hydrocarbons

PERIODICAL: V sb.: "Okisleniye uglevodorodov v zhidkoy faze", Moscow, AN SSSR, 1959, pp. 145 - 151

TEXT: The spread of the phenomenon of double effect (catalyzing and inhibiting effects) of metals of variable valency which was established with the example of Cu-stearate at the oxidation of n-decane (RZhKhim, 1959, No. 24, # 85348) was studied under the same conditions with the addition of Mn-stearate. It is shown that the induction period is the greater the higher the salt concentration, and that Mn-stearate has, analogous to Cu-stearate, catalyzing and inhibiting functions in the oxidation reaction of n-decane. The inhibiting function of metal stearates was not observed in the oxidation of isodecane. Z. Mayzus

Translator's note: This is the full translation of the original Russian abstract.
Card 1/1

86408

5.1190

2209, 1208, 1274

S/062/60/000/008/015/033/XX
B013/B055

AUTHORS: Vasil'yev, R. F., Kozlova, Z. G., ~~Chuchukina, L. G.~~,
Shlyapintokh, V. Ya., and Emanuel', N. M.

TITLE: On the Change in Catalytic Activity of Nickel Stearate
During the Oxidation of Ethyl Benzene

PERIODICAL: Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh nauk,
1960, No. 8, pp. 1337-1341

TEXT: The present publication treats a phenomenon observed during the nickel-distearate catalyzed oxidation of various hydrocarbons. The authors observed that in these reactions the maximum concentration of the hydroperoxide fairly equals its concentration in an uncatalyzed reaction. It was shown that the anomalous course of the kinetic curve of the hydroperoxide during the oxidation of ethyl benzene is connected with an inactivation of the catalyst. Various experiments were made to establish the cause of the reduced activity of the catalyst during the oxidation process (Figs. 3, 4). These experiments lead the authors to assume that products reacting with the catalyst and reducing its activity are formed during

Card 1/3

86408

On the Change in Catalytic Activity of Nickel S/062/60/000/008/015/033/XX
Stearate During the Oxidation of Ethyl Benzene B013/B055

the reaction. Since acids accumulate during the oxidation of the decomposition products of hydroperoxide, it seems likely that these very acids inactivate the catalyst, e.g. by forming insoluble salts (Refs. 2-4). Experiments performed in this direction showed that the reduced activity of the catalyst is indeed related to its reaction with these acids (Fig. 5). The established reduction of catalyst activity during the reaction permits a simple explanation for the accumulation of peroxides during the nickel-stearate catalyzed reaction (Figs. 1, 2). Till the maximum peroxide concentration is reached, the nickel salt is completely inactivated. The reaction is then practically uncatalyzed and the maximum peroxide concentrations are therefore in agreement. At the same time the maximum concentration is reached more quickly in the presence of nickel stearate since the latter has a strong catalytic effect at the outset of the reaction. The results of this investigation furnish further proof that in the catalytic oxidation of hydrocarbons metal salts are no catalysts but rather initiators of the process. Their activity, and frequently also the mechanism of their effect, change during the process. The observed reaction kinetics therefore reflect not only the properties of the reacting system, but also the changes in the activity and action of the catalyst in the

X

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On the Change in Catalytic Activity of Nickel Stearate During the Oxidation of Ethyl Benzene

86/08/60/000/008/015/033/XX
B013/B055

individual stages of the reaction. In studies of the catalytic mechanism, stabilization of the catalyst is particularly important. This would considerably facilitate the explanation of the mechanism of the catalytic effect of metal salts. There are 6 figures and 4 references: 3 Soviet and 1 British.

ASSOCIATION: Institut khimicheskoy fiziki Akademii nauk SSSR
(Institute of Chemical Physics of the Academy of Sciences
USSR)

SUBMITTED: February 18, 1959

Card 3/3

CHUCHULEV, Angel, inzh.

Supplying Sofia with pure, tasty, and high-mountain water from
the "Planshtitsa" reservoir. Khidrotekh i melior 9 no.6:180-
182 '64.

CHUCHULEV, Angel, inzh.

Possibilities of constructing the Sandanska Bistritsa Hydroelectric
Power System. Khidrotekh i melior 9 no.10:293-295 '64.

NEROBINA, N. (Nal'chik); CHUCHULIN, P. (Nal'chik)

Organizing extracurricular work on local geography by means of
radio. Geog. v shkole 25 no.4:55-58 J1-Ag '62. (MIRA 15:8)
(Kabardino-Balkar A.S.S.R.--Geography--Study and teaching)
(Radio in education)

CHUCHULIN, P.P.; YERMOLAYEV, A., ofitser-topograf zapasa (g.Ul'yanovsk);
PETRENKO, V.V. (g.Odessa)

Problems requiring discussion. Geog.v shkole 22 no.3:76-80
My-Je '59. (MIRA 12:11)

1. Kabardino-Balkarskaya ASSR (for Chuchulin).
(Geography--Study and teaching)

CHUCHUMOV, M. A. , Cand Med Sci -- (diss) "Chronaximetric studies of ~~the~~ vestibular and ^{visual} ~~optical~~ analyzers in certain mental diseases." Ivanovo, 1958. 11 pp (Ivanovo State Med Inst. Chair of Psychiatry). 200 copies.
(KL, 12-58,103)

-103-

L 10072-07 TET(n) DS/34
ACC NR: AP6029926 (A)

SOURCE CODE: UR/0413/66/000/0015/0089/0090

INVENTORS: Kolosnikov, G. S.; Tovlina, A. S.; Chuchun, A. Yo.; Barabashkina, I. A.; Yushmanova, V. A.

ORG: none

TITLE: Method for obtaining porous sulfo-ion-exchange resin. Class 39, No. 1844507 /announced by Moscow Institute of Chemical Technology imeni D. I. Mendeleev (Moskovskiy khimiko-tekhnologicheskii institut)

SOURCE: Izobret prom obraz tov zn, no. 15, 1966, 89-90

TOPIC TAGS: ion exchange resin, polymerization, porosity, polymer, resin

ABSTRACT: This Author Certificate presents a method for obtaining a porous sulfo-ion-exchange resin by graft copolymerization of styrol and a polymer containing isopropyl groups in the presence of a free-radical type initiator and of divinyl benzene as the cross-linking agent. The polymerization is followed by sulfonation with either sulfuric acid or weak oleum. To obtain a polymer with different porosity (capable of sorbing large organic ions), polyarylenealkyl is used as the isopropyl-group-containing polymer.

SUB CODE: 11/ SUBM DATE: 05Feb65

Card 1/1

UDC: 661.183.123.2:62-405.8:678.746.22-139:66.094.403

PONOMARENKO, A.A.; LITVINENKO, S.P.; SOLOV'YEVA, T.Ye.; CHUCHUPAK, V.D.

Chemiluminescence method for investigating the mixing and flow
of fluids. Dop. ta pov. L'viv. un. no.5 pt.2:88-89 '55.

(MLRA 9:10)

(Luminescence) (Hydrodynamics)

PONOMARENKO, A.A.; LITVINENKO, S.P.; SOLOV'YEVA, T.Ye.; GHIUCHIPAK, V.I.

Chemiluminescence method for investigating the mixing and flow of
liquids. Zav.lab.22 no.7:832-833 '56. (MLRA 9:12)

(Fluid dynamics) (Luminescence)

CHUCHUPAKA, K. D.

Chuchupaka, K. D.

"Geodetic Work in the Complex Mechanization of Earth Work on Large Hydraulic Constructions (The Example of Constructing the Kakhovka Hydroelectric Power Plant)." Min Higher Education USSR. Khar'kov Order of Labor Red Banner Agricultural Inst imeni V. V. Dokuchayev. Chair of Geodesy. Khar'kov, 1955 (Dissertation for the degree of Candidate in Technical Sciences)

SO: Knizhnaya letopis' No. 27, 2 July 1955

CHUCHUPALOV, V.A.

Paste-preparing machine for poultry and swine. Mekh. sil'. hosp.
12 no. 2:25 F '61. (MIRA 14:4)

1. Glavnyy inzh. sovkhoza No.8 im. Ordzhonikidze, Stalinskoy obl.
(Poultry—Feeding and feeds) (Swine—Feeding and feeds)

CHUCHUPALOV, V.A.

Electric hot water heating with thermostat control for dwellings.
Mekh. zil'. hosp. 12 no.9:25 S '61. (MIRA 14:11)

1. Glavnyy inzh. opytno-pokazatel'nogo sovkhoza im.
Ordzhonikidze, Stalinskoy oblasti.
(Hot-water heating)

MIKHAYLOV, P.G., kand.tekhn.nauk; CHUCHUSHKOV, M.K., inzh.; KUZ'KIN, V.A., inzh.

Increasing the efficiency of the system of working inclined layers
with filling. Sbor. KusNIUI no.9:20-42 '61. (MIRA 16:5)
(Kuznetsk Basin--Coal mine and mining) (Mine filling)

L 19755-65 EPA(s)-2/ET(m)/EP(t)/EP(b) Pt-10 IJP(c)/AEDC(b)/SSD/SSD(c)/AFWL/
 ASD(a)-5/RAEM(i)/RAEM(j)/ESD(gs)/ESD(t) JD/JG/MLK
 S/0000/64/000/000/0085/0087
 ACCESSION NR: AT5000424

AUTHOR: Lazebnaya, G.V., Romova, M.G., Chuchuyeva, R.

TITLE: Increasing the sensitivity of the flame-photometric determination of rubidium in cesium salts

SOURCE: Sibirskoye soveshchaniye po spektroskopii. 1st, Kemerovo, 1962. Spektroskopiya; metody* i primeneniye (Spectroscopy; methods and application). Doklady* soveshchaniya. Moscow, Izd-vo Nauka, 1964, 85-87

TOPIC TAGS: spectroscopy, flame photometry, rubidium determination

ABSTRACT: Using flame photometry, the authors determined rubidium in high-purity cesium chloride and cesium nitrate. The emission intensity of rubidium in the flame was increased 60-70% by the addition of 10 vol. % ethyl alcohol to the cesium salt solution; this made it possible to determine 0.001-0.0008% rubidium in the dry cesium salt. The behavior of the analytical lines of rubidium at 7800-7948 Å upon the addition of sodium chloride and ethyl alcohol was analyzed. On the basis of this study, the determination of rubidium was carried out by using the 7800 Å line. The method was checked by introducing known amounts of rubidium. The sensitivity achieved, $0.8-1 \times 10^{-3}$, is not the maximum attainable value. The authors suggest the use of certain

Card 1/2

L 19755-65

ACCESSION NR: AT5000424

instruments which will raise the sensitivity still further. Orig. art. has: 2 figures and
1 table. 0

ASSOCIATION: none

SUBMITTED: 09 May64 ENCL: 00

SUB CODE: GC

NO REF SOV: 002 OTHER: 004

Card 2/2

SKLYAR, V.T., kand.khim.nauk; SABIROVA, G.V., kand.khim.nauk; ZHURBA,
A.S., kand.khim.nauk; ROZHIN, V.P., inzh.; GONOPOL'SKIY, L.Ye.,
inzh.; ZVEREVA, A.D., inzh.; CHUCHVARA, P.G., inzh.; Prinsipali
uchastiy: KOVAL'CHUK, L.V.; TERENT'YEVA, V.N.; VEDERNIKOVA, V.T.

Production of the RPF-12 freon oil from Anastas'yevka petroleum.
Nauch.zap.Ukrainiproekta no.8:48-57 '62. (MIRA 16:1)
(Freons) (Lvov--Petroleum--Refining)

SABIROVA, G.V.; MAN'KOVSKAYA, N.K.; PORUTSKIY, V.P.; TEREENT'YEVA, V.N.; KOVAL'CHUK,
L.V.; LEBEDEVA, L.B.; ROZHIN, V.P.; GONOPOL'SKIY, L.Ye.; CHUCHVARA, P.G.

Studying petroleum growth-promoting substances in the petroleum re-
fineries of the Ukraine. Nefteper. i neftekhim. no.7:13-16 '64.

1. UkrNIIGIproneft' i L'vovskiy neftepererabatyvayushchiy zavod. (MIRA 17:11)

JEDLICKA, Jaroslav; ~~CHUDACEK~~, Frantisek

Carbon tetrachloride poisoning. Vnitr. lek., Brno 1 no.5:
339-344 May 55.

1. Z vnitřní kliniky, přednosta doc. MUDr. Karel Bobek a z
oddělení pro choroby z povolání KUNZ, přednosta MUDr.
Frantisek Chudacek, v Plzni, Praha XIII, Tr. SNB 694.

(CARBON TETRACHLORIDE, poisoning
diag.,)

(POISONING
carbon tetrachloride, diag.)

CHUDACEK, Frant., MUDr.; SRUTEK, Josef, MUDr.

Method of work hygiene and prevention of occupational disease
at the health center in the Lenin Works in Plzen. Pracovní lek.
8 no.5:369-372 Oct 56.

1. Zavodni ustav narodniho zdravi pri Leninovych zavodech v
Plzni, reditel MUDr. Fr. Chudacek.
(INDUSTRIAL HYGIENE,
health centers in indust. in Czech. (Cz))

21(3,4)

PHASE I BOOK EXPLOITATION

CZECH/2404

Habanec, V., Doctor; J. Havelka, Engineer; Zd. Hlasivec,
Doctor of Medicine; Zb. Hrdlička, Engineer; I. Chudáček
(Graduate in Physics); V. Kouřim, Engineer; J. Kuba,
Doctor of Natural Sciences; V. Myslivec, Professor; Jan
Tůma, Engineer; and M. Voříšek (Graduate in Physics)

Atom a jaderná technika (The Atom and Nuclear Engineering)
Praha, Naše vojsko, 1957. 290 p. (Series: Universita
vojáka) 4,000 copies printed.

Reviewers: Bittner, Engineer; Drška, Engineer; Hrdlička,
Engineer; Kulka, Engineer; Spurný, Doctor; and Šimáně,
Engineer; Ed.: Stanislav Vobořil.

PURPOSE: The book is intended for the general reader.

COVERAGE: The book outlines the principles and operation of
nuclear power plants and the use of radioisotopes. The intro-
ductory chapters cover the fundamentals of nuclear physics and
radioactivity. Several subsequent chapters deal with reactor
physics, types of reactors, their engineering, control and
Card 1/12

The Atom and Nuclear Engineering

CZECH/2404

instrumentation. Operating and planned nuclear power installations are described. A short chapter is devoted to the possibility of using nuclear power in transportation. The remaining chapters report on radioisotopes for industry, and on radiology, radiation hazards and safety measures. No personalities are mentioned. There are 25 references, all Czech.

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AVAILABLE: Library of Congress

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TM/ec
10-12-59

CZECHOSLOVAKIA/Nuclear Physics - Installations and Instruments. C
Methods of Measurement and Research

Abs Jour : Ref Zhur Fizika, No 10, 1959, 21934

Author : Chudacek, Ivo

Inst : Charles University, Prague, Czechoslovakia

Title : Energy Spectrum of Alpha Particles Emitted from Sources
of Different Thicknesses

Orig Pub : Chekhosl. fiz. zh., 1958, 8, No 4, 396-403

Abstract : An investigation was made of the energy distribution of
alpha particles emitted by sources of different thick-
nesses. A semi-empirical formula is obtained for the
energy distribution $N(E)dE$ and is in satisfactory
agreement with the results of the experiment. The sour-
ce of alpha particles is powdered U_3O_8 , from which layers
of various thicknesses were obtained by precipitation

Card 1/2

CZECHOSLOVAKIA/Nuclear Physics - Installations and Instruments. C
Methods of Measurement and Research

Abs Jour : Ref Zhur Fizika, No 10, 1959, 21934

in ether. The spectrum of the alpha particles was measured in nuclear emulsion "Agfa" K2 (200 microns). The nucleon emulsion was superimposed directly on the U_3O_8 film and exposed for 12 hours. The spectrum of five sources was measured with thicknesses from 0.15 to 70 mg/cm². The total number of the measured tracks of alpha particles amounted to 3592. The formula obtained for the energy distribution can be used also for alpha particles emitted from other sources, for example, from Ra. --
N.I. Petukhova

Card 2/2

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CZECHOSLOVAKIA/Nuclear Physics - Installations and Instruments. C
Methods of Measurement and Research

Abs Jour : Ref Zhur Fizika, No 11, 1959, 24374
Author : Drskay Ladislav; Chudacek, Ivo; Sterba, Frantisek
Inst : Charles University, Prague, Czechoslovakia
Title : Measurements of Certain Neutron Spectra by the Method
of Nuclear Emulsions
Orig Pub : Ceskosl. casop. fys., 1958, 8, No 5, 589-598
Abstract : Using tracks of recoil protons in a nuclear emulsion,
the author has determined experimentally the energy
spectrum of the neutrons from a Ra + Be source and the
reaction Li (d,n) on a thick target. A comparison is
made of the results obtained with the data of other
authors. The spectrum of neutrons from a simplified model
of Ra + Be system is calculated theoretically for

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Measurement of some neutron spectra by the nuclear emulsion method. L. Drška, I. Chudáček, and F. Štěpánek (Karlova Univ., Prague). *Czechoslovak J. Phys.* 8, 648-67 (1953) (in Russian).—The neutron spectrum is measured of a Ra-Be source. The source contained 50 mg. Ra as RaCl_2 . The shape of the frequency vs. energy diagram of the neutron spectra is dependent on the exptl. arrangement and layer thickness. Another part deals with the $\text{Li}^7(d,n)\text{Be}^8$ reaction; the shape of the energy spectrum and the energy levels of a Be^8 nucleus are investigated. An energy level of 5.4 m.e.v. is found in addn. to already known ones. 32 references. A. Kremheller

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CZECHOSLOVAKIA/Nuclear Physics - Installations and Instruments. C-2
Methods of Measurement and Research

Abs Jour : Ref Zhur - Fizika, No 4, 1959, No 7476

Author : Chudacek Ivo
Inst : Karlov University, Prague, Czechoslovakia
Title : Energy Spectrum of Alpha Particles from Sources of Various Thickness

Orig Pub : Ceskosl. casop. fys., 1958, 8, No 3, 319-325

Abstract : The energy spectrum of the alpha particles emitted by sources of various thicknesses, made of U_2O_8 , were investigated. The spectrum was measured with an Alfa K2 nuclear emulsion 200 microns thick. The actual energy spectrum $N(E)dE$ can be approximated by the semi-empirical expression

$$N(E)dE = K(E_0^{3/2} - E^{3/2})^2 E^M dE$$

(where E_0 is the initial particle energy, and K and M are certain constants) which describes well the observed spectrum. -- V.I. Lond'yol

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Distr: 4E3c/4E3d

Energy spectrum of α -particles emitted from sources of different thicknesses. Jvo Chudáček (Karlova Univ. Prague). *Czechoslov. J. Phys.* 6, 396-403 (1958) (in English).
—The energy distribution of α -particles emitted from a U_3O_8 source is measured. The energy distribution $N(E)dE$ found can be expressed satisfactorily by a semi-empirical formula which is based on Geiger's formula (Segré, *Experimental Nuclear Physics*, 1953, 211 (C.A. 47, 0288i)) for α -particle energies from 0 to 8 m.e.v. Altogether 3802 trajectories of α -particles are investigated in 200- μ thick nuclear emulsion ACPA K2 of 5 plates.
A. Kremheller

C.H.U.D.A.C.E.K., I.

CZ/37-58-5-10/19

AUTHORS: Dráka, L., Chudáček, I. and Štěpán, J.

TITLE: Measuring of Certain Neutron Spectra by Means of the Method of Nuclear Emulsions (Měření některých neutronových spektrů metodou jaderných emulzí).

PERIODICAL: Československý časopis pro fyziku, 1958, Nr. 5, pp 589-598 (Czech)

ABSTRACT: As a part of systematic work aimed at solving certain problems of fast neutron physics, the energy spectrum was measured by means of frequently used sources of fast neutrons. The method of nuclear emulsions is studied in the first part of the paper. The spectrum is analyzed at an $Ra + Be$ source and its characteristic is entered in the neutron spectrum of the $Ra + Be$ source is entered in the graph, Fig. 1. In Fig. 2 the neutron spectra compared. $Ra + Be$ sources measured by various authors are compared. In Fig. 3 the applied approximation of the spectrum of the $Ra + Be$ source reacting with beryllium is graphed and also the characteristic of the product of the neutron spectrum the theoretical characteristic of the measured spectrum of an $Ra + Be$ source is graphed. The measured spectrum of the neutron spectrum of an $Ra + Be$ source is in good agreement with the results obtained by other authors and a

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satisfactory explanation of this spectrum is given in the paper. In the second part of the paper the spectrum is studied of neutrons from a thick lithium target which is bombarded with deuterons. The measurements were made for three sets of test conditions, the basic characteristic of which are summarized in Table 1. In Fig. 5 part of the corrected spectrum of neutrons is graphed for the test arrangement A and a neutron energy of $E_n > 5$ MeV. In Fig. 6 the results are graphed of measurements of the neutron spectrum for the arrangement C, Table 1, for the energy range E_n between 1 and 11 MeV. The obtained results are in satisfactory agreement with results obtained by other authors (Refs 28-35). From analysis of the spectra, the energy levels were arranged in the spectrum of the $Ra + Be$ source, which are partly consistent with current data. The obtained results are in agreement with some more recent measured data. Acknowledgments are made to Prof. Dr. V. Petržílka

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for suggesting the subject of this work. There are 6 figures, 5 tables and 41 references, 1 of which is Czech, 2 Soviet, 4 German, 1 Hungarian, 1 Swiss and 32 English.

ASSOCIATION: Fakulta technická a jaderná fyzika Karlovy university, Praha (Faculty of Technical and Nuclear Physics, Charles University, Prague)

SUBMITTED: January 14, 1958

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CHUDACEK, I.

CZECHOSLOVAKIA/Nuclear Physics - Installation and Instrument.
Methods of Measurement and Research.

C

Abs Jour : Ref Zhur Fizika, No 1, 1960, 349
Author : Drska, L., Chudacek, I., Sterba, F.
Inst :
Title : Measurement of Certain Neutron Spectra by the Method
of Nuclear Emulsions
Orig Pub : Chekhosl. fiz. zh., 1958, 8, No 6, 648-657
Abstract : See Referat Zhur Fizika, No 11, 1959, 24374.

Card 1/1

CHUDACEK, I.; SODOMKA, L.

The influence of pressure on the luminescence of zinc sulfide
single crystals. Chekhosl fiz zhurnal 13 no.3:209-210 '63.

1. Fakulta technicke fyziky, Liberec.

CHUDACEK, I.

Approximative formula of the Luminous intensity of
piezoluminescence. Acta physica Pol 26 no.3/4:599-
604 S-O '64.

1. College of Mechanical Engineering, Liberec, Czechoslovakia.

L 21436-66 EMP(t) IIP(c) M
ACC NR: AP5013937

SOURCE CODE: CZ/0055/65/015/005/0359/0362

AUTHOR: Chudacek, I.

ORG: Engineering and Textile College, Liberec

TITLE: Periodic triboluminescence of zinc sulfide ²¹ 27

SOURCE: Chekhoslovatskiy fizicheskiy zhurnal, v. 15, no. 5, 1965, 359-362

TOPIC TAGS: zinc sulfide, triboluminescence, material deformation, luminescent material, solid mechanical property, potential energy, dielectric property, potential energy, dielectric property, thermal excitation, light pulse

ABSTRACT: An investigation of the periodic mechanical excitation of the triboluminescence of ZnS was carried out. It was proved experimentally that periodic triboluminescence can be exerted by periodically supplying deformation energy to a ZnS luminophore. This effect occurs for different types of ZnS luminophores (electroluminophore, photoluminophore) and different activators. The strongest brightnesses occur with an electroluminophore activated by Mn. At high pressures, triboluminescence does not exhibit periodicity. It was found that the periodicity of triboluminescence disappears at a pressure of 700 kg/cm². At lower pressure, the triboluminescence brightness exhibits periodicity, but it is limited in time. This led to the conclusion that even for small pressures, triboluminescence is an irreversible phenomenon of a destructive nature. The creation of triboluminescence

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does not depend on the medium in which the excited luminophore occurs. This was verified on resins and nitrolacquer. The time dependence of triboluminescence depends on the mechanical properties of the surrounding medium. With solid substances (resin), the transfer of mechanical energy is faster. With soft substances (lacquer), the transfer of mechanical energy is slower. The form of triboluminescent light pulses is thus connected with the mechanical properties of the binding agent in which the ZnS is suspended. The conclusion that triboluminescence is a phenomenon of a destructive nature agrees with the work of Stranski and others (G. Wolff, I. Schönewald, I. N. Stranski, Zeit. Kristallogr. Mineralog. Petrog. 106, 1954), who observed the triboluminescence of substances only when the latter were crushed. The second conclusion that triboluminescence does not depend on the dielectric properties of the luminophore's surroundings was expressed in a paper (G. Alzetta, N. Minaya, S. Santucci, Nuovo Cim. XXIII, 1962, 910). In the latter case, the binding agents were various liquids. Orig. art. has: 3 figures. [Based on author's abstract.]

[NT]

SUB CODE: 11, 20/ SUBM DATE: 17Jun64/ OTH REF: 005/

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CHUDACEK, J.; MOLEK, J.; CHYBA, E.

Production of high-pressure vessels. p. 208.

STROJIRENSKA VYROBA. (Ministerstvo tazkeho strojirenstvi, Ministerstvo presneho strojirenstvi a Ministerstvo automobiloveho prumyslu a zemedelskych stroju) Praha, Czechoslovakia. Vol. 7, no. 5, May 1959.

Monthly list of East European Accessions (EEAI), LC, Vol. 8, no. 10, Oct. 1959. Uncl.